

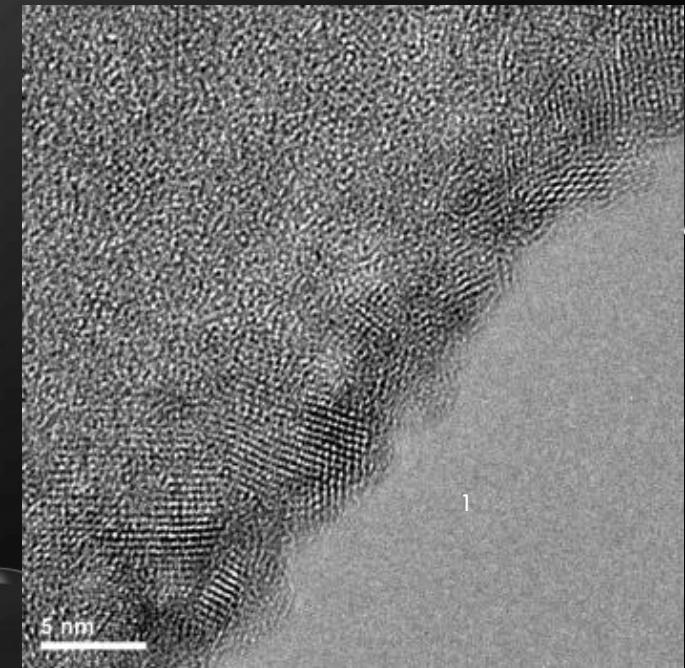
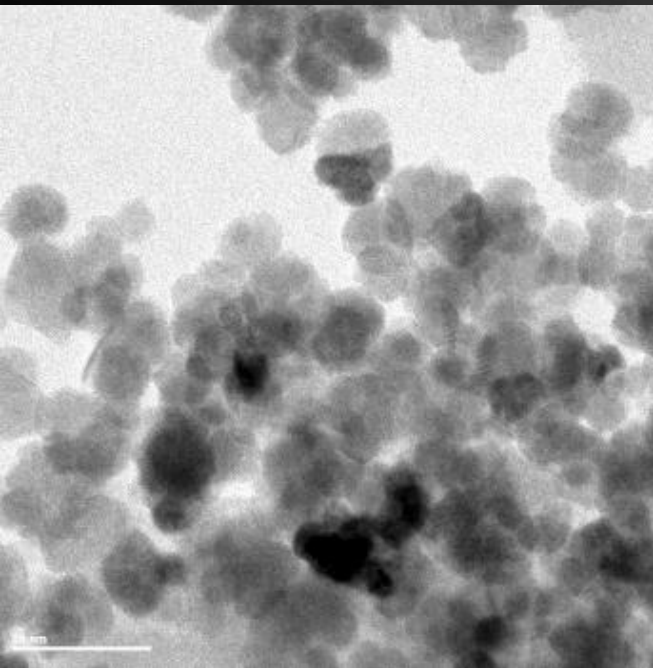
# NANOTECHNOLOGY: AN EMERGING AND PROMISING TECHNOLOGY FOR WASTEWATER TREATMENT

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# Outline

- Why nanotechnology (NT)?
- NT for wastewater in the context of Kingdom
- Nano for different treatments
- **Applications:**
  - Industrial dye degradation by  $\text{TiO}_2$
  - Antimicrobial properties of nanoparticles (NPs)
- Challenges in NT
- Societal and health impact
- Acknowledgement

# Why nanotechnology (NT)?

High  
surface  
area

Possible to  
tune the  
properties of  
the  
nanoparticles

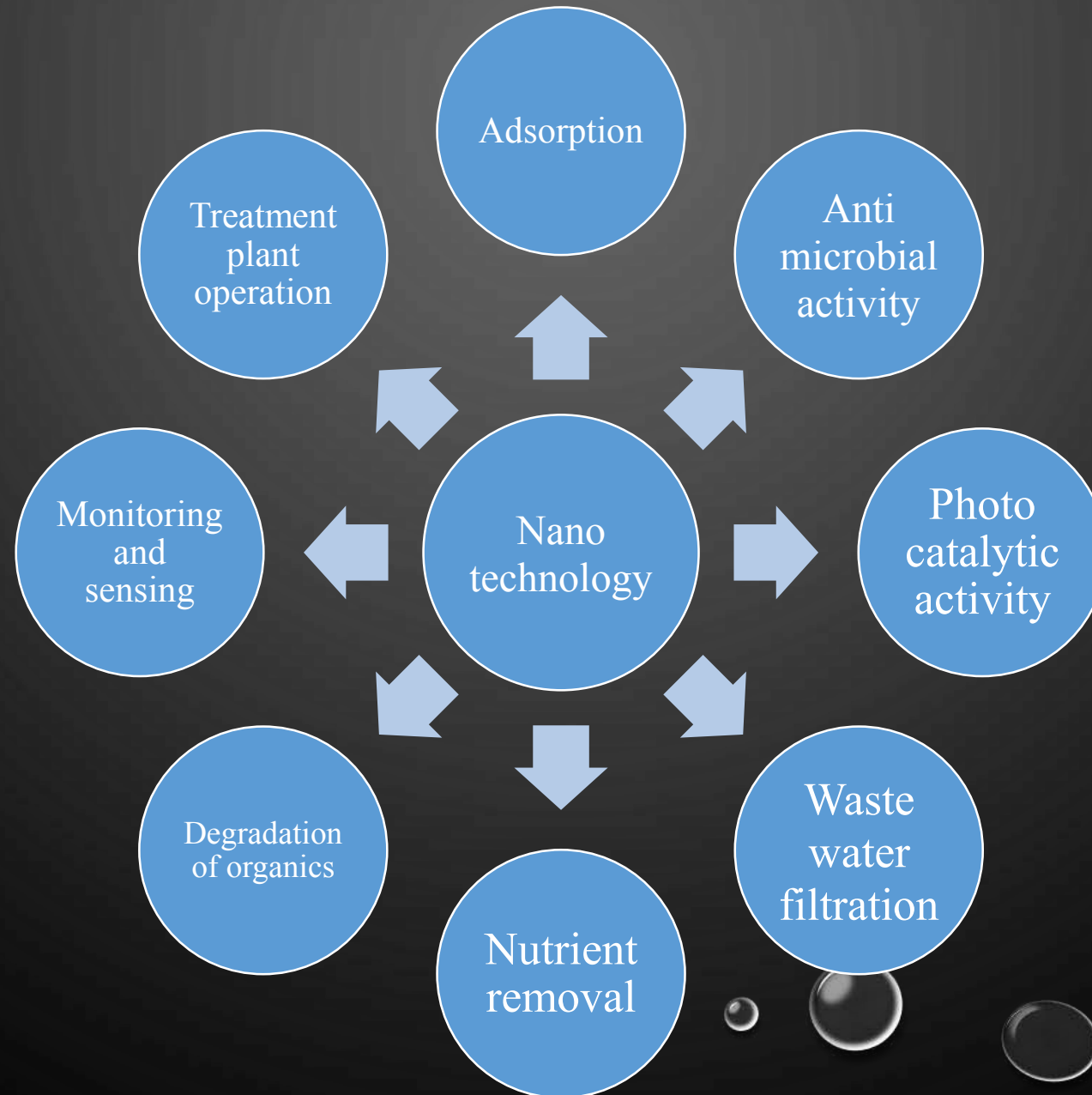
Can increase  
competitive-  
ness among  
the  
wastewater  
treatment  
technologies

Environmental  
performance  
can be  
improved by  
increasing  
efficiency and  
reducing  
pollutants

Can take the  
lead in  
wastewater  
treatment  
technologies  
by deducing  
the  
knowledge  
gap

Cost-  
effective-  
ness

# NT for wastewater in the context of Kingdom



# Nano for different treatments

Applications	Nanoelements
Adsorption	TiO <sub>2</sub> ; CNT; Fe <sup>0</sup> ;
Antimicrobial activity	TiO <sub>2</sub> ; MgO; ZnO; nAg; CNT; Chitosan; Fe <sup>0</sup> ; Al <sub>2</sub> O <sub>3</sub> ; Fullerene;
Photocatalytic activity	TiO <sub>2</sub> ; ZnO; CNT; Fullerene;
Wastewater filtration	TiO <sub>2</sub> ; CNT; Fe <sup>0</sup> ; nAg; Graphene;
Nutrient removal	TiO <sub>2</sub> ; Fe <sup>0</sup> ;
Degradation of organic	TiO <sub>2</sub> ; Fe <sup>0</sup> ; CNT;
Monitoring and sensing	CNT; Quantum dots; nAu;
Treatment plant operations	TiO <sub>2</sub> ; ZnO; CNT;

## Applications:

- Industrial dye degradation by  $\text{TiO}_2$
- Antimicrobial properties of nanoparticles (NPs)



# Why TiO<sub>2</sub> for industrial dye degradation?

Low  
cost

Non-  
toxic

Particle  
stability

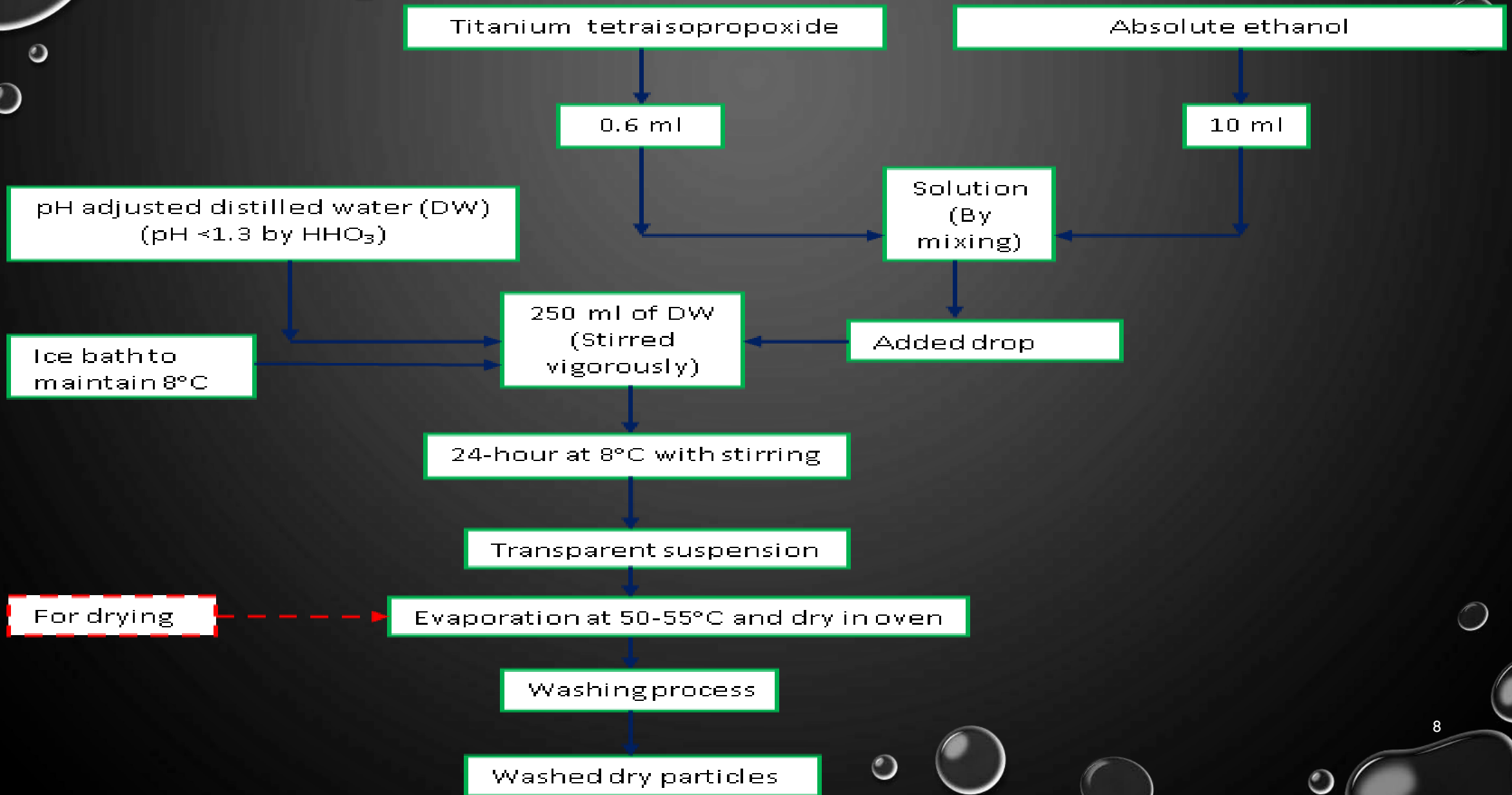
Anti-  
microbial  
properties

Environmental  
friendliness

Heavy  
metal  
adsorption

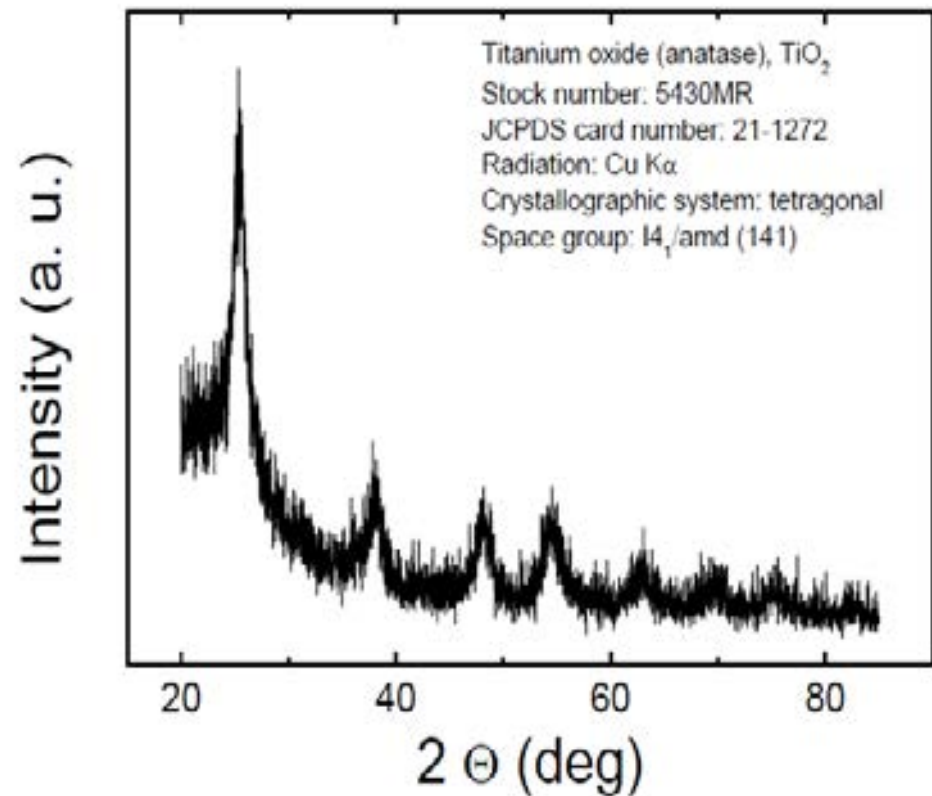
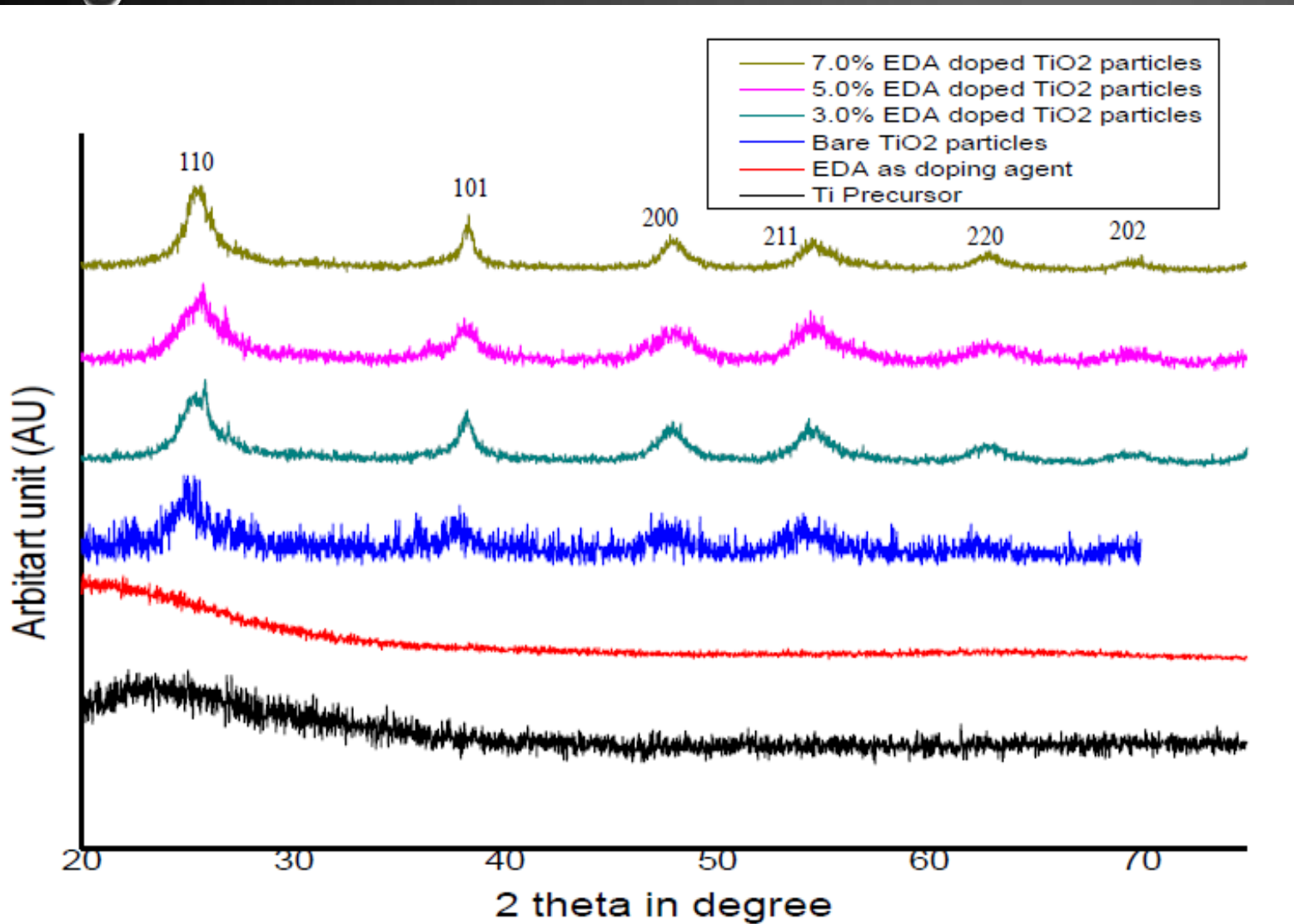
Generate  
ROS to  
inactivate  
virus &  
bacteria

# TiO<sub>2</sub> synthesis for Photocatalytic activity

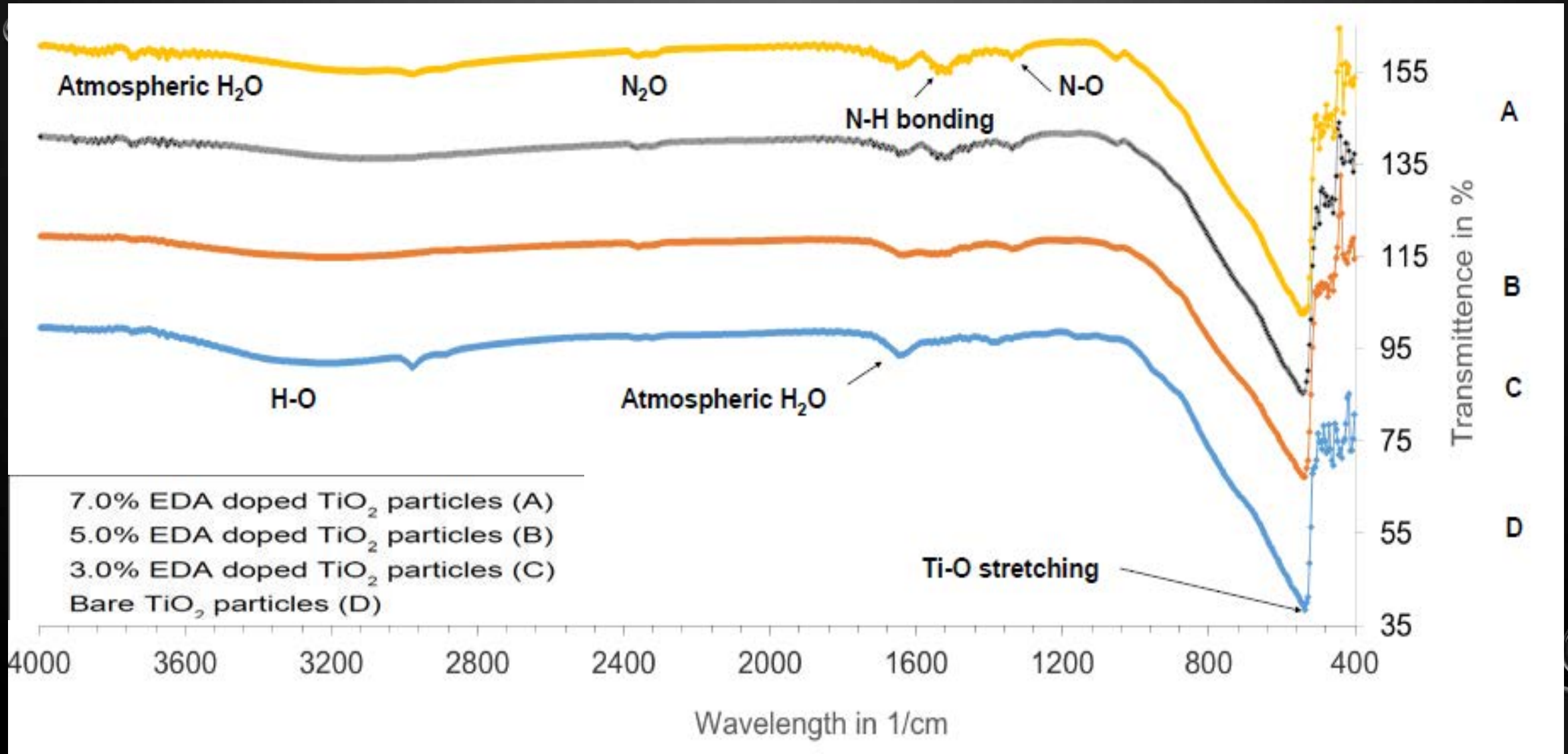




# XRD pattern for synthesized TiO<sub>2</sub> Anatase (2 $\theta$ at 25.5; 38.5; 48.2; 54.4; 62.4)



# FT-IR Spectra for synthesized TiO<sub>2</sub> (N-H bonding around 1500/cm)



# Industrial dye degradation by TiO<sub>2</sub>

EDA concentration (% v/v)	MB photo-degradation efficiency
Bare TiO <sub>2</sub> particles	58.14%
3.0% EDA-TiO <sub>2</sub>	92.88%
5.0% EDA-TiO <sub>2</sub>	83.68%
7.0% EDA-TiO <sub>2</sub>	92.70%

# Antimicrobial activity by nanoparticles (NPs)

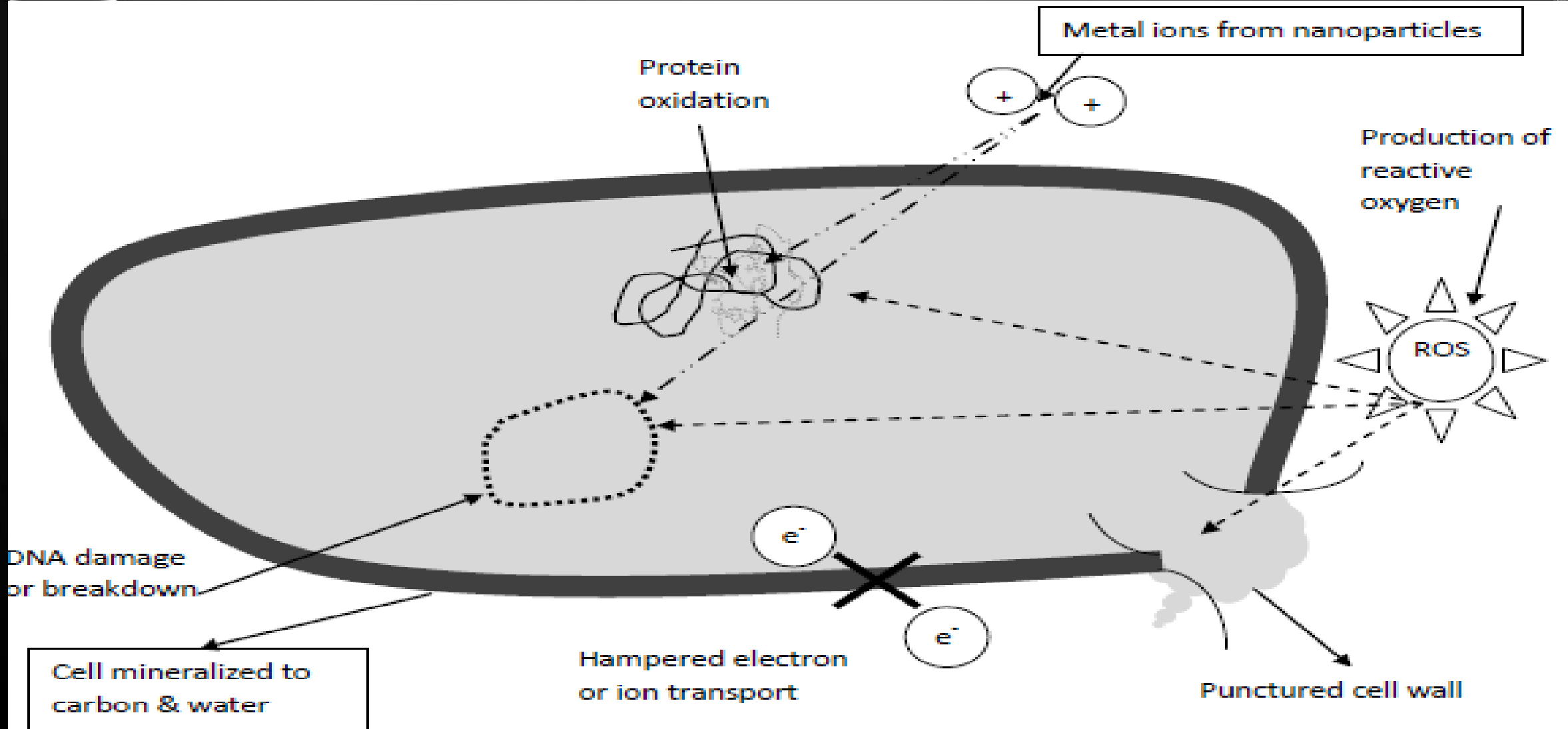


Figure 1: Bacteria cell damage by generated ROS (adapted from Li et al. 2008; Huh & Kwon, 2011)



# Antimicrobial activity by NPs

**Table 1**  
The efficiency of TiO<sub>2</sub> NPs as antimicrobial agent.

Elements	Properties	Irradiation process	Matrix	Inactivation efficiency	Ref.
TiO <sub>2</sub> powder (Degussa P-25)	Surface area 60 m <sup>2</sup> /g	Solar light (1660 μE s <sup>-1</sup> m <sup>-2</sup> ) & dark	Drinking water	<i>E. coli</i> (99.97% in light after 60 min; 99.96% in dark after 60 min) Initial TiO <sub>2</sub> conc. 1.0 g/l & initial <i>E. coli</i> conc. 10 <sup>6</sup> CFU/ml	Wei et al. (1994)
TiO <sub>2</sub> -Fe <sub>2</sub> O <sub>3</sub> composite	Particle size 10 nm; Surface area 168 m <sup>2</sup> /g;	UV (11 W; Wavelength 253.7 nm)	Drinking water (pH-7.0)	<i>E. coli</i> (99.9% after 1 min.) Initial TiO <sub>2</sub> conc. 0.2 g/l & initial <i>E. coli</i> was 10 <sup>7</sup> CFU/100 ml	Sun et al. (2003)
TiO <sub>2</sub> powder (Degussa P-25)	Surface area 60 m <sup>2</sup> /g	UV (18 W; Wavelength 300-420 nm; 7.9 × 10 <sup>-6</sup> einsteins/l/s)	Drinking water (pH-7.1)	MS2 Phage, a virus species (88.78% after 120-min) & <i>E. coli</i> (99.43% after 120-min.) Initial TiO <sub>2</sub> conc. 1.0 g/l & initial <i>E. coli</i> conc. 6.4 × 10 <sup>5</sup> CFU/ml; MS2 same.	Cho et al. (2005)
TiO <sub>2</sub> film	Particle size 8-10 nm; Surface area 147 m <sup>2</sup> /g	UV (2 × 15 W; Wavelength 365 nm; 3.48 mW/cm <sup>2</sup> ) & Dark	Wastewater	MC-LR (100% after 150-minute; 24% after 180-minute); <i>E. coli</i> (100% after 120-minute) Initial TiO <sub>2</sub> conc. 62.2 μg/cm <sup>2</sup> of film & initial <i>E. coli</i> conc. 10 <sup>6</sup> -10 <sup>7</sup> CFU/ml	Choi et al. (2006)
TiO <sub>2</sub> & N-doped TiO <sub>2</sub> (Doped by Ethylendiamine & Ethanolamine)	Particle size 10 nm; Surface area 80-120 m <sup>2</sup> /g;	Solar light (10 W/m <sup>2</sup> )	Synthetic wastewater (pH-7.2)	<i>E. coli</i> (99.99% by TiO <sub>2</sub> after 120-min.; 100% by TiO <sub>2</sub> -N-Ethylendiamine after 120-min.; 99.9999% by TiO <sub>2</sub> -N-Ethanolamine after 120-min.) Initial TiO <sub>2</sub> conc. 0.1 g/l & initial <i>E. coli</i> conc. 8.9 × 10 <sup>8</sup> CFU/ml	Liu et al. (2007)

# Antimicrobial activity by NPs

**Table 2**  
Antimicrobial efficiency of nAg, CNT and ZVINP.

Elements	Properties	Matrix	Inactivation efficiency	Ref.
Ag-Zeolite		Water	<i>E. coli</i> (99.7% after 30 min); Ag-Zeolite conc. 100 mg/ml and <i>E. coli</i> conc. $2.2 \times 10^7$ CFU/ml	Matsumura et al. (2003)
n-Ag	Particle size 12 nm; Surface area 158 m <sup>2</sup> /g	Water	<i>E. coli</i> (99.0% after 30 min); n-Ag conc. 50 µg/ml and <i>E. coli</i> conc. 10 <sup>8</sup> CFU/ml	Sondi and Sondi (2004)
n-Ag		Water	<i>E. coli</i> (99.93% after 24 h); <i>B. subtilis</i> (99.99% after 24 h); n-Ag conc. 100 µg/ml	Yoon et al. (2007)
SWNT	5 µm	Water	<i>S. typhimurium</i> (99.997% after 1 h); SWNT conc. 100 µg/ml and <i>S. typhimurium</i> cell 10 <sup>8</sup> CFU/ml	Yang et al. (2010)
SWNT		Water	<i>E. coli</i> K12 (73.1 ± 5.4%, 79.9 ± 9.8% and 87.6 ± 4.7% inactivation after 30, 60 and 120 min of incubation time, respectively)	Kang et al. (2007)
SWNT and MWNT		Water	<i>E. coli</i> K12 (79.9% and 25% after 1 h by SWNT and MWNT respectively); NT conc. 5 µg/ml and <i>E. coli</i> conc. $5 \times 10^7$ CFU/ml	Kang et al. (2008)
SWNT and MWNT	SWNT (Diameter 1.2 nm; Length 17.8 µm); MWNT (Diameter 17.4 nm; Length 77 µm)	Water	<i>E. coli</i> (87% after 1 h); <i>P. aeruginosa</i> (80% after 1 h); <i>S. epidermidis</i> (66% after 1 h); <i>B. subtilis</i> (16% after 1 h); NT conc. 0.1 µg/ml and Bacteria conc. $2 \times 10^6$ CFU/ml	Kang et al. (2009)



# Challenges in NT

- Particle agglomeration
- Reducing particle surface area
- Bulk commercial production
- Practical applications
- Utilization of exceptional physical, chemical and mechanical properties
- NPs may have less toxicity than micro or bulk size particles

# Societal and health impact

- Nanotoxicity: About 400 reported toxicity and ecotoxicity
- Public awareness
- Nanocontamination
- Can effect human cells
- Recovering of nanoparticles or nanomaterials from the environment

# Acknowledgement





Thank you

Question!!